

Total number of printed pages-7

1 (Sem-5/FYUGP) CHE02MJ

2025

CHEMISTRY

(Major)

Paper : CHE0500204

(Organic Chemistry-II)

Full Marks : 45

Time : 2 hours

The figures in the margin indicate full marks for the questions.

1. Answer the following questions as directed :
1×5=5

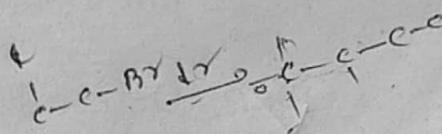
(a) Isopropyl bromide on Wurtz reaction gives

(i) n-Hexane

(ii) Propane

(iii) 2,3-Dimethylbutane

(iv) None of the above



B14F0 0013

Contd.



(b) Which of the following is laevorotatory ?

- (i) Glucose
- (ii) Fructose
- (iii) Sucrose
- (iv) Cellulose

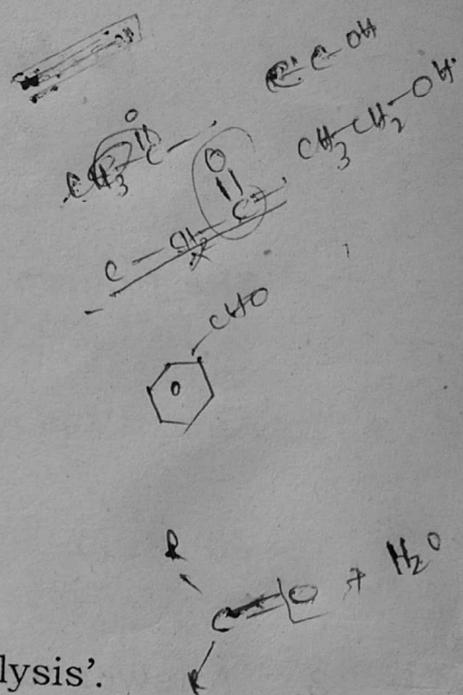
(c) Which of the following will not undergo aldol condensation reaction ?

- (i) Propanal
- (ii) Ethanal
- (iii) Propanone
- (iv) Benzaldehyde

(d) Menthol is a

- (i) Monoterpene
- (ii) Diterpene
- (iii) Sesquiterpene
- (iv) Triterpene

(e) Define 'Ketonic hydrolysis'.



2. Answer **any five** of the following questions :
2×5=10

(a) What is sulphur ylide? What are the different types of sulphur ylide ?

- (b) What is meant by hydroboration oxidation? Illustrate it with an example.
- (c) How will you synthesise *n*-butanoic acid from ethyl acetoacetate?
- (d) What is Cannizzaro reaction? Give one example.
- (e) Give *Re* or *Si* nomenclature to two faces of acetophenone.
- (f) What is Cram's rule? Give an example.
- (g) *D*-fructose is a ketohexose, yet it reduces Fehling's solution. Explain.
- (h) Why do glucose and mannose form same osazones?
- (i) Give the structures of isomeric citrals.
- (j) Suggest a mechanism for the acid catalysed mutarotation of *D*-(+)-glucose.

3. Answer **any four** from the following questions : 5×4=20

- (a) What is the difference between stereoselectivity and stereospecificity in chemical reactions? Give examples of each. 3+2=5

(b) ✓ What is the Wurtz reaction? Give an example with mechanism. What are the drawbacks of Wurtz reaction?

1+2+2=5

(c) How does the Felkin-Anh model differ from the Cram's rule? Provide an example where the two models predict different outcomes.

3+2=5

(d) ✓ How are phosphines converted to phosphonium salts and phosphorous ylides? Show *one* synthetic use of triphenylphosphine.

2+3=5

(e) ✓ What do you mean by active methylene compound? Write the reaction for preparation of ethyl acetoacetate. Provide a mechanism for the reaction.

1+1+3=5

(f) ✓ "Citral is an acyclic monoterpene." Explain this by writing the characteristics of such compounds. How many stereoisomers of citral are possible? How these stereoisomers can be distinguished? Write a synthesis of citral.

1+1+1+2=5

(g) Draw the cyclic anomeric forms of D-fructose. Give the mechanism for hydrolysis of glycoside under acidic condition.

1+4=5

(h) Write notes on : $2\frac{1}{2} \times 2 = 5$

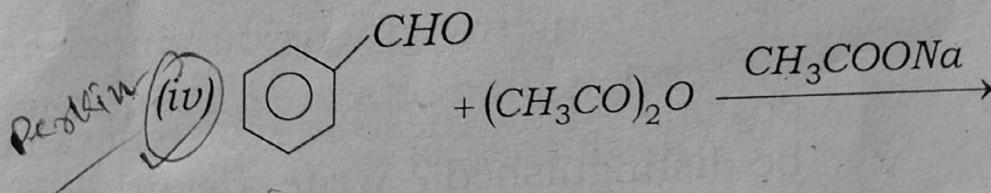
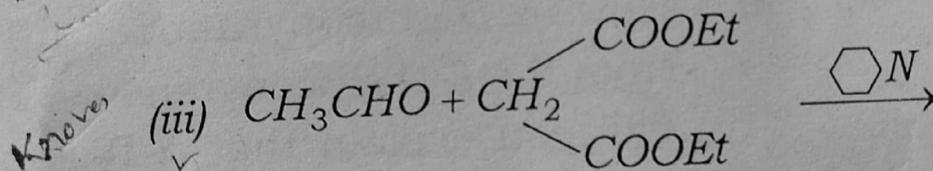
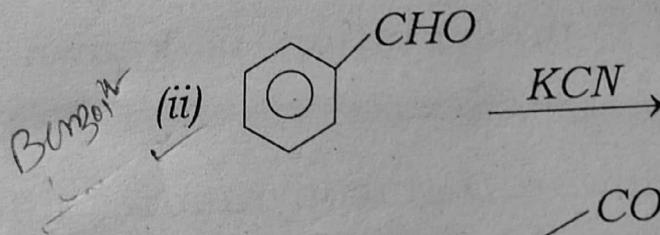
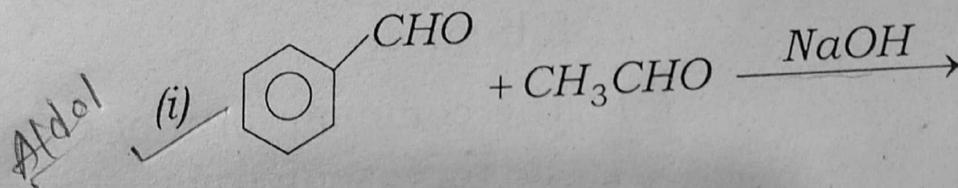
(i) Formation of stability of enamines

(ii) Alkylation of enolates

4. Answer **any one** of the following :

$10 \times 1 = 10$

(a) Complete the following reactions and give suitable mechanisms for each of them : $2\frac{1}{2} \times 4 = 10$



(b) (i) Explain why polysaccharide do not mutarotate. 2

(ii) Give the structures of sucrose, lactose and maltose. 3

(iii) Fill in the blanks : $1 \times 5 = 5$

(A) *D*-glucose is an epimer of *c*-*L*.

(B) Ketoses have less number of _____ than aldoses.

(C) Mild oxidation of glucose gives gluconic acid.

(D) _____ is present mostly as furanose.

(E) The common form of glucose as represented by Haworth presentation is known as _____.

(c) (i) How α -*D*-glucopyranose differs from the β -*D*-glucopyranose? How will you explain that α -*D*-glucose reduces Fehling's solution but α -*D*-methylglucoside does not? What products are obtained when *D*-glucose is—

(A) treated with excess of $PhNHNH_2$ followed by dilute hydrochloric acid;

(B) treated with periodic acid? n-hexon

$1+2+2=5$

(ii) Define the following reactions and give the mechanisms : $2\frac{1}{2} \times 2 = 5$

(A) Beckmann rearrangement

(B) Wolff rearrangement

(d) (i) How will you convert an aldopentose to an aldohexose by using Kiliani synthesis? Explain each step by using suitable chemical reactions. 5

(ii) How will you prepare the following compounds from diethylmalonate? $2\frac{1}{2} \times 2 = 5$

(A) Succinic acid

(B) Crotonic acid

